

reSPECT

PET & SPECT

Principles of PET & SPECT similar.

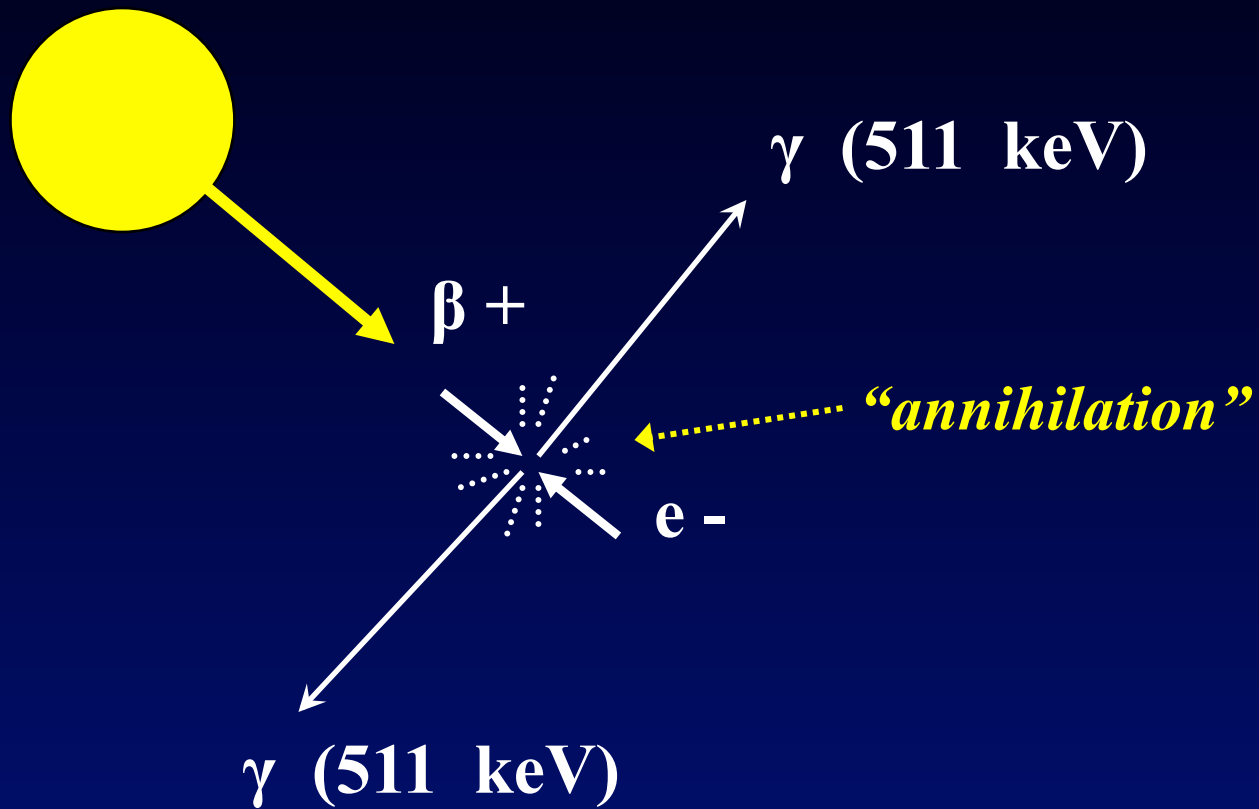
PET & SPECT differ in

- radiochemistry
- instrumentation
- physics of photon emission

Decay Characteristics of Commonly Used PET & SPECT Nuclides

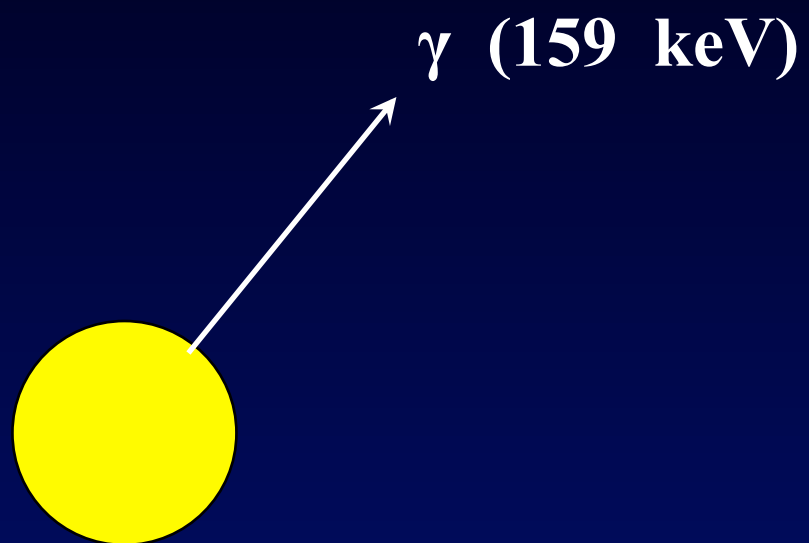
<u>Radionuclide</u>	<u>T_{1/2}</u>	<u>Photon Energy</u>
PET		
15-Oxygen (¹⁵ O)	2.1 min	511 keV
13-Nitrogen (¹³ N)	10.0 min	511 keV
11-Carbon (¹¹ C)	20.3 min	511 keV
18-Fluorine (¹⁸ F)	109 min	511 keV
SPECT		
99mTechnetium (^{99m} Tc)	6 h	140 keV
123Iodine (¹²³ I)	13 h	159 keV
133 Xenon (¹³³ Xe)	5.3 d	80 keV

PET





SPECT

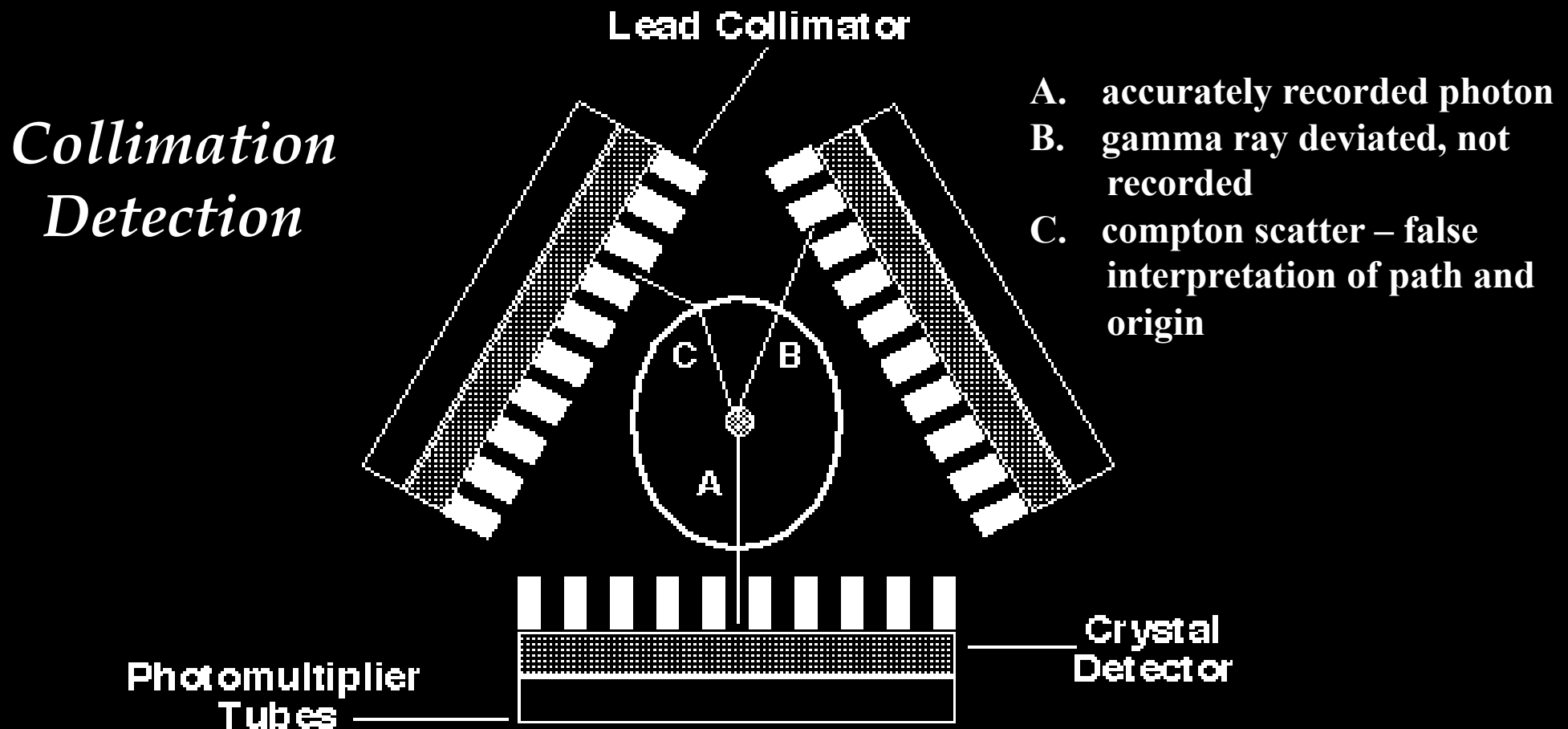




SPECT Scanner

[Adapted from Malison et al., Psychopharmacology: The Fourth Generation of Progress

Ed Bloom ED & Kupfer DJ, NY p865-879 1995]

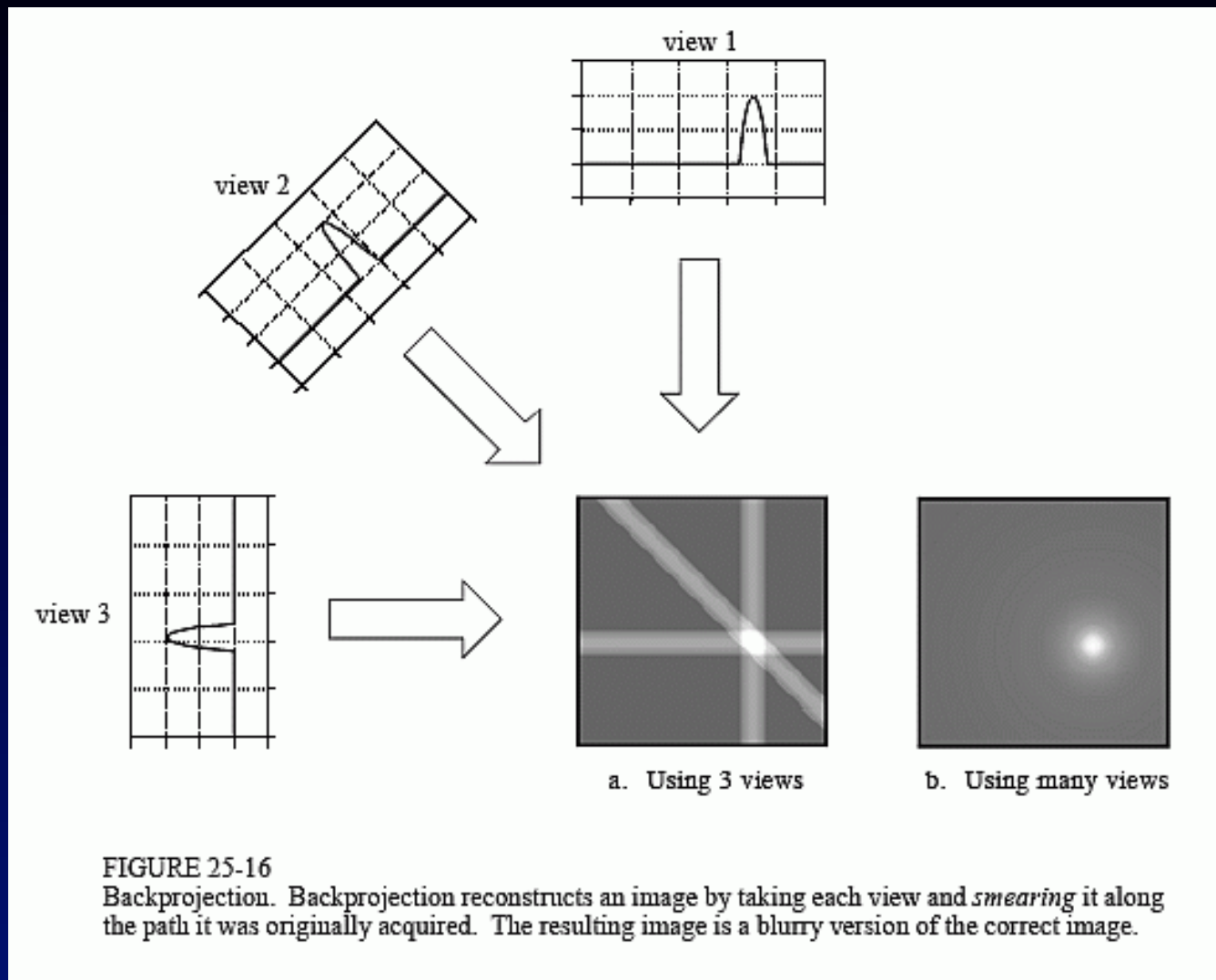


SPECT: Collimation Detection

Collimator-Typically a block of lead with drilled holes placed between the source of activity and the camera's sodium iodide crystal detectors

- Photon energy is absorbed by crystal, light in the visible range is emitted and is detected by photomultiplier tubes
- Holes in collimator set limits on the possible locations of the source of emission because the walls (septae) block the photons that enter at too sharp of an angle
- By viewing the source of activity from multiple angles, the reconstruction algorithm uses the principle of backprojection to determine the distribution of activity in the source

Filtered Back-Projection



PET vs SPECT

- radioisotope $T^{1/2} < 2$ h
- requires rapid synthesis by robot or radiochemist
- requires onsite cyclotron
 - costs \$1-2.5 million
 - Technical staff for operation & maintenance
- $\beta^+ + e^- \rightarrow$ two 511 keV photons
- Coincidence detection
- radioisotope $T^{1/2} > 6$ h
- allows more time for synthesis of radiopharmaceutical
- produced commercially (^{123}I) or locally onsite very inexpensively ($^{99\text{m}}\text{Tc}$)
- single γ photon 159 keV
- Collimation detection